## JOURNAL OF THE CHEMICAL SOCIETY ERRATA

Vol. 1961, page 67, Table 5, col. $k_{1}$. For $3.2 \times 10^{9}$ read $3.2 \times 10^{3}$.
Vol. 1964, page 5899, Table 1, col. x/a. For N(1) . . . $0 \cdot 0393$ read 0.3093 .
Vol. 1964, page 3950 , line 5. For hydroxylamine hydrochloride ( 0.7 g ., 0.01 mole ) and triethylamine ( $1.0 \mathrm{~g} ., 0.01 \mathrm{~mole}$ ) read hydroxylamine hydrochloride ( $1.4 \mathrm{~g} ., 0.02 \mathrm{~mole}$ ) and triethylamine ( 1.8 g ., 0.018 mole).

Vol. 1964, page 5892. Formula (IV) should be shown as


Vol. 1964, page 6059, equation (8). For $p=[p \tilde{M}(1-z) / M(2-z)]^{(1-z)}$ read $p=[M p(1-z) /$ $\overline{\mathrm{M}}(2-z)]^{(1-2)}$.

Vol. 1964, page 6080 , line 9 . Insert $\rightleftharpoons$ in the space between the L.H. and R.H. side of the equation.
Vol. 1964, page 6137, line $8^{*}$. For 18 hr . read 4 days.
Vol. 1964, page 6138, line 1. For 4 days read 18 hr .
Vol. 1964, page 6256, Paper 1199, line 1. For up to $90^{\circ}$ read up to $190^{\circ}$. Formula (I) should be shown as


Vol. 1965, page 1034, line $7^{*}$. For $\beta-\mathrm{H}$ configuration read $\alpha-\mathrm{H}$ configuration. \{J. W. Clark-Lewis [Rev. Pure Appl. Chem. (Australia), 1962, 12, 96] has obtained optical rotatory dispersion evidence which shows that demethylbromopterocarpin has $\alpha(\mathrm{H})$-configuration at $\mathrm{C}_{3}$ and $\left.\mathrm{C}_{4} \cdot\right\}$

Vol. 1965, pages 133 and 134. The last three lines on p. 133 should be transposed with the first two lines of print on page 134. The paragraph will then read:

This mechanism, or an elaboration implicating the 17-hydroxy-group, does not offer a complete explanation of the hydrolysis of the 16,21-diacetate (VII; $R=R^{\prime}=A c$ ), with 1 equivalent of base, since neither of the mono-esters (VII; $R=A c, R^{\prime}=H$ or $R=H, R^{\prime}=A c$ ) nor a 17,21-ortho-ester would be expected to hydrolyse spontaneously to the triol. ${ }^{12}$

Vol. 1965, page 577, line 18. For 2,3,4,5-Tetrafuoro-4-hydroxypyridine read $2,3,5,6$-Tetrafluoro-4hydroxypyridine.

Vol. 1965, page 752, Table, column labelled Radical. For $\cdot \mathrm{O}-\mathrm{NM}-\mathrm{CO}_{2}^{-}$read $\cdot \mathrm{O}-\mathrm{NH}-\mathrm{CO}_{2}^{-}$; for $\left(\cdot \mathrm{O}-\mathrm{NM}-\mathrm{SO}_{3}\right)_{-}$read $\left(\cdot \mathrm{O}-\mathrm{NH}-\mathrm{SO}_{3}\right)^{-}$; for $(\cdot \mathrm{O}-\mathrm{NM}-\mathrm{CO}-\mathrm{Ph})$ read $(\cdot \mathrm{O}-\mathrm{NH}-\mathrm{CO}-\mathrm{Ph})$.

Vol. 1965, page 1224, Paper 215, line 11. For has shown most of the promising activity read has shown the most promising activity.

## Errata

Vol. 1965, page 1429. Formula (IV) should be shown as


Vol. 1965, page 1447, line 3. For $\beta=\left\{z R_{1}\left[1-1 /\left(2^{2}\right)+1 /\left(2^{2} D_{0}\right)-1 / D_{8}\right]\right\} /\left[R_{\mathrm{x}}\left(\frac{1}{2}+\frac{1}{2} D_{0}-1 / D_{8}\right)\right]$ $\left.\operatorname{read} \beta=\left\{z R_{1}\left[1-1 /\left(2^{z}\right)+1 /\left(2^{z D_{0}}\right)-1 / D_{8}\right]\right\} /\left\{R_{\mathrm{x}}\left[\frac{1}{2}+1 /\left(2^{D_{0}}\right)-1 / D_{8}\right)\right]\right\}$

Vol. 1965, page 2271, line 6. For 2-ethoxycarbonyl-t-butylamine hydrochloride read 2 -carboxyethyl-t-butylamine hydrochloride.

Vol. 1965, page 2979 , Paper 534. In all names, for naphtho[ $\left.2^{\prime}, 1^{\prime}-b\right]$ pyran read naphtho $\left(2^{\prime}, 1^{\prime}: 2,3\right)$ pyran.
Vol. 1965, page 3194, Paper 584, line 5. For The length of the Co-O bond is $1.99 \AA$. The five Co-N bond-lengths vary between 1.94 and $1.93 \AA$. read The length of the Co-O bond is $1.93 \AA$. The five $\mathrm{Co}-\mathrm{N}$ bond-lengths vary between 1.94 and $1.99 \AA$.

Vol. 1965, page 3198, Table 4. Under Angle, for $\cdots \mathrm{N}\left(3^{\left.3^{\mathrm{iv}}\right)} 84\right.$ read $\mathrm{N}\left(3^{\mathrm{ix}}\right) 84$.
Vol. 1965, page 949, Table 1, column 7-H. For (II) insert $J_{7,8}=7$; For (XXIII), for $\mathbf{3 7 \cdot 4}(\mathrm{t})$ read $\mathbf{3 \cdot 4 7 ( t )}$; column 9a-H, for XIX, for $J_{\mathrm{Me}_{2} \mathrm{CH}}=8$ read $J_{\mathrm{MeCH}_{2}}=8$.

Vol. 1965, page 2634, line 13. For derivative (VII) read derivative (VIII).
Vol. 1965, page 3201, Table 2. Entry (XIV), $\mathrm{CHCl}_{3}$. Insert after $6 \mathrm{H}, 4.61: J_{5,6}=6.0 \pm 0.3$.
Vol. 1965, page 3206, line 16. After 2-methylbenzothiazole, insert ( 10 g .).
Vol. 1965, page 3378 , line 28 . For $\mathrm{F}, 31 \cdot 8 \%$ read $\mathrm{F}, 48 \cdot 7 \%$.
Vol. 1965, page 3902, Paper 725. Structure (II) should appear as


Vol. 1965, page 3915, in structure (XX). For $\mathrm{SO}_{2} \mathrm{Me}$ read SOMe.
Vol. 1965, page 3917. For $N N^{\prime}$-(isopropylideneamino)guanidine read $N N^{\prime}$-di(isopropylideneamino)guanidine.

Vol. 1965, page 4008. Structure (IV) should read


Vol. 1965, page 4289, line 8. For 1-methyl-2-vinylcyclopropane read 1-methyl-2-methylenecyclopropane.
Vol. 1965, page 4513, Table, reference $\dagger$. For $N$-methyl groups at C-3 and C-20, read $N$-methyl groups at $\mathrm{C}-20$ and $\mathrm{C}-3$.

Vol. 1965, page 4520 , in structure (XXX). The H atom on carbon- 13 should be replaced by a methyl group.

Vol. 1965, page 4590, Table. Column headed Formula. For $\mathrm{C}_{20} \mathrm{H}_{11} \mathrm{BrNO}$ read $\mathrm{C}_{20} \mathrm{H}_{12} \mathrm{BrNO}$, and for $\mathrm{C}_{20} \mathrm{H}_{12} \mathrm{Br}_{2} \mathrm{NO}$ read $\mathrm{C}_{20} \mathrm{H}_{11} \mathrm{Br}_{2} \mathrm{NO}$.

Vol. 1965, page 4956, Table 2. Last two lines of the second column:
For $\quad * N$-methyl-2-pyrrolidone ( $o$-Xylene) 0 read $N$-methyl-2-pyrrolidone (o-Xylene)

## Errata

Vol. 1965, page 5561, line 22*. For $p$-anisidine salt read p-anisidine salt.
At end of paragraph, add: (Found: $\mathrm{C}, 46 \cdot 6 ; \mathrm{H}, 4 \cdot 9 ; \mathrm{N}, 14 \cdot 7 . \mathrm{C}_{11} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{6}$ requires $\mathrm{C}, 46 \cdot 6 ; \mathrm{H}$, $4 \cdot 6 ; \mathrm{N}, 14 \cdot 8 \%$ ).

Vol. 1965, page 5746 , line $15^{*}$. For 4.5 g . read 8.9 g .

Vol. 1965, page 5747, line 20. For 1 -acetylanthracene ( $99 \%$ ) read 1 -acetylanthracene ( $9 \%$ ).
Vol. 1965, page 6072, Table 1, Column 6, " Eluant." For System II read System III.
Vol. 1965, page 6548, line 22. For $\mathbf{7 2} \%$ of 4 -hydroxycyclohexyl phosphate read $\mathbf{7 2} \%$ of the related diol diphosphate.

Journal: Index of Authors, 1964.
Papers by Sir Ewart R. H. Jones and Emrys R. H. Jones have been confused.
Sir Ewart R. H. Jones's Papers appear on pp. 1173, 1161, 1476, 3210.
Emrys R. H. Jones's Papers appear on pp. 5907, 5911, 5919, and 5922.

* From bottom of main text.

